

Published on Web 05/12/2009

## Full Color Stop Bands in Hybrid Organic/Inorganic Block Copolymer Photonic Gels by Swelling–Freezing

Changjoon Kang,<sup>†</sup> Eunjoo Kim,<sup>†</sup> Heeyoel Baek,<sup>†</sup> Kyosung Hwang,<sup>†</sup> Dongwoo Kwak,<sup>†</sup> Youngjong Kang,<sup>\*,†</sup> and Edwin L. Thomas<sup>‡</sup>

Department of Chemistry, Hanyang University, 17 Haengdang-dong, Seongdong-gu, Seoul 133-791, Korea, and Department of Materials Science and Engineering, Massachusetts Institute of Technology, 77 Massachusetts Avenue, Cambridge, Massachusetts 022139

Received March 19, 2009; E-mail: youngjkang@hanyang.ac.kr

Block copolymers have been widely investigated for fabricating functional nanomaterials due to their unique properties of self-assembly.<sup>1-4</sup> For example, block copolymers have been used as platforms for creating various 1D, 2D, and 3D photonic crystals, periodic dielectric structures where the propagation of electromagnetic waves is prohibited over a certain range of frequencies.<sup>5–13</sup> Typically, block copolymer photonic crystals exhibiting stop bands in the visible region of wavelengths have employed extremely highmolecular-weight polymers ( $\sim 10^6$  g/mol) with narrow molecular weight distribution. Hence, tailoring the optical properties of block copolymer photonic crystals is limited by the difficulty of synthesis and processing of such high-molecular-weight copolymers. To address this problem, Kang et al. recently reported swollen photonic gels, which demonstrated that selective swelling of the block copolymer lamellar structure allows extremely large tunability of the photonic stop band from the UV region to the IR region ( $\lambda_{peak}$ = 350-1600 nm).<sup>14</sup> In this case, lower-molecular-weight block copolymers ( $< 5 \times 10^4$  g/mol) can also be used as materials for creating photonic crystals by swelling, but obviously the optical properties of the swollen photonic gels disappear when the gels dry.

Herein, we report a facile way of fabricating organic/inorganic hybrid 1D photonic crystals exhibiting stop bands in the visible regime after drying by self-assembly of the modest-molecularweight gel block copolymers and a subsequent stop band fixation process we term "swelling-freezing" (Scheme 1). The general idea

## Scheme 1



is that the easily tunable photonic gels in solution phase can be used as a template for incorporation of components that will transform the soft gel to a solid. To demonstrate our idea, swollen photonic gels were first prepared by self-assembly of polystyrene*block*-poly(2-vinylpyridine) diblock copolymer PS<sub>190</sub>-*b*-P2VP<sub>190</sub> ( $M_n$ × 10<sup>3</sup> = 190/190, PDI = 1.10, purchased from Polymer Source Inc., Dorval, Canada) and then subsequently fixed with SiO<sub>2</sub>. PS<sub>190</sub>*b*-P2VP<sub>190</sub> photonic gel films were prepared as previously described.<sup>14</sup> Briefly, in-plane lamellar films were prepared by spincasting from a 5% PS<sub>190</sub>-b-P2VP<sub>190</sub> solution in propylene glycol monomethyl ether acetate and subsequently annealing the film in chloroform vapor at 50 °C for 7 days. The as-prepared, dry films were transparent since both the periodicity and the refractive index contrast were insufficient to exhibit stop bands in the visible region of wavelengths. The photonic gel films were then swollen with methanol, a selective solvent for the P2VP blocks, and displayed a uniform blue color. Spectrophotometer data show two stop bands at 465 and 928 nm (Figure S1, Supporting Information). Since the primary band is out of the visible region, the visible blue color is due to the second-order peak, positioned at 465 nm. On the basis of the UV/vis spectrum, we calculated the size of swollen P2VP microdomains using the transfer matrix method,15 which gave  $t_{\rm P2VP}$ (swollen)  $\approx 280$  nm – almost 5.5 times the original domain size  $(t_{P2VP} = 50 \text{ nm})$ .<sup>14</sup> This suggests that the swollen P2VP domains can be filled with other materials since the content of P2VP in the layer is only 18% by volume. Infiltration of SiO<sub>2</sub> was carried out by sol-gel processing using tetraethoxysilane (TEOS) as a precursor for SiO<sub>2</sub>.<sup>16-18</sup> TEOS was added to the methanol bath containing the swollen gel ([TEOS] = 0.45 M), and then ammonium hydroxide was gradually added with vigorous stirring ( $[NH_4OH] = 0.15 M$ ), which made the solution basic with pH = 11. After 20 min of stirring at room temperature, the pH of solution was lowered to pH = 9 by adding concentrated hydrochloric acid. Upon addition of hydrochloric acid, the clear solution gradually became turbid and viscous, and eventually formed gel-like solids. The films were aged at pH = 9 for different periods of time ( $\tau_a$ ) to complete the fixation. The final frozen photonic gels ("f-photonic gels") were obtained after drying the films in air. In basic condition, TEOS is catalytically hydrolyzed, forming discrete colloidal particles, and they grow with time. These colloidal particles aggregate into a continuous network, forming a SiO<sub>2</sub> gel under lowered pH.<sup>19</sup> When the sol-gel reaction was carried out entirely either at pH = 11 or pH = 9, the fixation failed not only because of insufficient capability of infiltration but also because of formation of a thick residual coating over the film which deteriorates the optical properties by random scattering. We found that, by sequentially lowering the pH from 11 to 9, the structure of swollen photonic gels could be appropriately fixed.

After drying, the *f*-photonic gels showed strong reflective colors for months without any change. The persistent reflective color of the dry *f*-photonic gels suggests that the swollen photonic gels were successfully fixed by infiltration of SiO<sub>2</sub>. Fixation was fully controllable and reproducible. As shown in Figure 1, the position of stop bands of the *f*-photonic gels could be easily varied with increasing aging time  $\tau_a$ . For a short aging time ( $\tau_a = 15$  min), a well-defined stop band first appeared at  $\lambda_{peak} = 381$  nm, gradually shifted to longer wavelengths as the aging time increased, and finally

<sup>&</sup>lt;sup>†</sup> Hanyang University. <sup>‡</sup> Massachusetts Institute of Technology.



Figure 1. (a) Transmission spectra of dry f-photonic gels prepared at different aging times ( $\tau_a = 15$  min to 2 h). (b) Variation in  $\lambda_{peak}$  with  $\tau_a$ . The line is provided as a guide to the eye only. (c) Color variation of the f-photonic gel films with aging time. Photographs were taken on a black background under white light illumination.

reached  $\lambda_{\text{peak}} = 741$  nm when the sample was aged for 2 h. The shift of  $\lambda_{peak}$  with aging time implies that the amount of SiO<sub>2</sub> infiltrated within the P2VP domains gradually increased, and accordingly the P2VP/SiO<sub>2</sub> domain size and the overall periodicity also increased. We found that the position of the stop band increased approximately logarithmically as a function of the aging time (Figure 1b). The bandwidth also gradually increased with increase of the aging time. Bandwidth broadening is mainly due to the increase of the refractive index contrast between PS ( $n_{\rm PS} \approx 1.6$ ) and P2VP ( $n_{\rm P2VP} \approx 1.6$ ) domains as more SiO<sub>2</sub> ( $n_{
m SiO_2} \approx$ 1.22-1.46)<sup>20</sup> is incorporated into the P2VP domains.<sup>14</sup> Samples with long aging time ( $\tau_a > 1$  h) showed another peak in the short wavelength region other than the primary peak. This peak can be attributed to the second-order peak, since it satisfies the relationship  $\lambda_{\text{first peak}} = 2\lambda_{\text{second peak}}^{21}$  The existence of the higher order peak indicates that the lamellar structures have good long-range order after the SiO<sub>2</sub> infiltration process.

Scanning electron microscopy (SEM) of the f-photonic gels confirmed that the swollen lamellar structures were successfully fixed by SiO<sub>2</sub> (Figure 2). Samples for cross-sectional imaging were



**Figure 2.** SEM images of *f*-photonic gels prepared at (a)  $\tau_a = 15$ , (b) 45 and (c) 120 min. Samples for SEM imaging were prepared by fracturing f-photonic gels immediately after freezing in liquid nitrogen. SEM images were taken after osmium coating. Inset cartoons show the P2VP/SiO<sub>2</sub> layers as thicker green layers and the glassy PS blocks as thin red layers. With increasing amounts of silica in the P2VP layers, the silica particles bridge across from the P2VP layers, forming a third layer comprised of silica and solvent (air after drying).

prepared by fracturing *f*-photonic gels immediately after freezing in liquid nitrogen. In this case, layers defining the PS and P2VP/ SiO<sub>2</sub> domains are revealed due to their different mechanical properties.

The thin, bright layers in the images are the PS domains, and the thicker, somewhat darker layers correspond to the P2VP/SiO<sub>2</sub> domains. As shown in Figure 2, the thickness of the P2VP/SiO<sub>2</sub> layers increases with aging time, while that of the glassy PS layers remains constant. At  $\tau_a = 15$  min, the periodicity, the distance between the two thin PS layers marked with arrows, was 110 nm, which is slightly larger than that of the original photonic gels (d =100 nm).<sup>14</sup> The periodicity increased to 210 nm and to 295 nm when the aging time increased to  $\tau_a = 45$  min and 2 h, respectively. These images directly prove that SiO<sub>2</sub> was infiltrated into the P2VP domains, further confirmed by energy dispersive X-ray spectroscopy analysis (Figure S5, Supporting Information). Interestingly, at higher loadings of silica ( $\tau_a > 45$  min), the originally interdigitated P2VP chains were split into two parts and subsequently induced void regions with silica bridges connecting P2VP layers. Currently we are investigating its mechanism.

In conclusion, we have demonstrated a facile method of fabricating robust photonic crystals from modest-molecular-weight block copolymers which takes advantage of a swelling and freezing process. Controlled infiltration of SiO<sub>2</sub> within the swollen P2VP domains allows fixating and tailoring the stop bands across the full visible portion of the spectrum (400-800 nm). We are currently investigating other f-photonic gels infiltrated with metal oxides, such as TiO<sub>2</sub>, ZnO, Al<sub>2</sub>O<sub>3</sub>, and ZrO<sub>2</sub>. We propose that this method could provide a simple avenue to fabricating hybrid organic/inorganic photonic crystals using block copolymers as a template.

Acknowledgment. This research was supported by a grant to Y.K. from Information Display R&D Center (F0004100-2008-31), one of the 21st Century Frontier R&D Program funded by the Ministry of Knowledge Economy of Korean government, and by a NSF grant to E.L.T. (DMR-0804449).

Supporting Information Available: Experimental details for preparation and fixation of photonic gels, UV-vis spectra for swollen photonic gels, SEM images for other f-photonic gels, and the results of EDS analysis. This material is available free of charge via the Internet at http://pubs.acs.org.

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- JA9021478